BAND AND TOTAL EMISSIVITY OF AMMONIA

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INFRARED radiation of ammonia has received considerable attention in the past because of its common presence in industry and in planetary atmospheres [1-6]. Port [1] measured total radiation from ammonia vapor-nitrogen mixtures at a total pressure of 1 atm and a pathlength of $51\cdot 2$ cm by use of a complex but ingeniously designed system involving a windowless high-temperature absorption cell. On the basis of these measurements in the range of consist of the v_2 band located around 950 cm⁻¹ (10.5 μ), the ($v_4 + 2v_2$) bands around 1628 cm⁻¹ (6.15 μ), and the ($v_1 + v_3 + 2v_4$) bands around 3300 cm⁻¹ (3 μ) [5]. The overtone bands, $2v_2$ and $2v_4$, are much weaker than their respective overlapping fundamental bands. Infrared bands between 20 and 35 μ have also been measured [6], but their contributions to the total radiation of ammonia in the temperature range of practical interest (300-1000°K) are

Band intensity $(cm^{-2} atm^{-1})$ Band center (cm^{-1}) Transition Degeneracy Band [5] (47 for $v_1 + v_3 + 2v_4$) 790 [3] [4] 20 0000-1000 $v_1(3.0 \ \mu)$ 3337 1 $v_2(10.5 \ \mu)$ 677 600 950 0000-0100 1 $(47 \text{ for } v_1 + v_3 + 2v_4)$ 0000-0010 2 13 $v_3(2.9 \ \mu)$ 3448

0000--0001

Table 1. Fundamental bands of ammonia

2

133

temperature from 300°K to 1000°K and ammonia partial pressure from 0.006 atm to 1 atm, a total-emissivity chart has been constructed [1, 2]. Recently, extensive measurements of infrared band absorption of ammonia have been made at 300°K by France and Williams [5]. The purpose of the present work is (1) to correlate the recent France-Williams band data to yield Edwards wide-band parameters [7, 8] and (2) to predict the total emissivity and to compare with the early measurements of Port.

1627

v4(6-15 µ)

negligible. Relevant information regarding the four fundamental bands is given in Table 1.

 $(150 \text{ for } v_4 + 2v_2)$

110

In accordance with the Edwards exponential wide-band model [7, 8], the France-Williams room temperature data can be correlated to yield the basic wide-band parameters as shown in Table 2. The correlation procedure as well as the definition of these parameters can be found in numerous previous investigations [7–10]. For extrapolation to higher temperatures, the formulas given by Edwards and Balakrishnan [8] are used. The total emissivity of ammonia

The major infrared band-absorption regions of ammonia

Table 2. Wide-band parameters of ammonia $(300^{\circ}K)$				
ameters	$v_1 + v_3 + 2v_4$	$v_4 + 2v_2$	v ₂	

Parameters	$v_1 + v_3 + 2v_4$	$v_4 + 2v_2$	v ₂
$C_1 (\text{cm}^{-2} \text{ atm}^{-1})$	47	150	790
C_{2} (cm ^{-$\frac{3}{2}$} atm ^{-$\frac{1}{2}$})	73	100	134
$C_{3}(cm^{-1})$	106	97.4	106-3
n	0.7	0.7	0.7
В	6.0	5.77	6.2



FIG. 1. Total emissivity of ammonia.

as calculated from this correlation and extrapolation procedure is shown in Fig. 1 in comparison with the early findings of Port. It should be noted that in Fig. 1 p_a stands for the ammonia partial pressure, and L for the geometric mean beam length. The agreement between the suggested values of Port and the present prediction is quite good indeed. It is a little surprising, however, to see that the agreement in the extrapolated (higher temperatures) region is better than that at 300°K, since the prediction at 300°K should be most reliable as it is purely based on the France-Williams band absorption data.

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HEAT TRANSFER PARAMETERS OF A PARALLEL PLATE HEAT EXCHANGER

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NOMENCLATURE

a, distance between the plates through which laminar flow occurs;

b, wall thickness;

 C_i , specific heat of fluid, i;

g, dimensionless velocity distribution of the laminar ---side fluid, u/\bar{u} ;

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